

Supporting Information

Solvent triggered shape morphism of 4D printed hydrogels

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1. Rheological characterization

In rheological testing, flow behaviour, shear thinning behaviour and angular frequency sweep of the hydrogels were investigated by varying the percentage of chitosan from 1-8%. Fig. S1a) revealed the flow behaviours of different hydrogels, in which sol-like 1-5% chitosan hydrogels exhibited a good gravity-induced flow. The 6% and 7% hydrogels remained stable after inverting the hydrogel-containing tubes for 5 min after which it flowed due to gravity. The 8% chitosan hydrogel exhibited a good gel-like behaviour and didn't flow even at 10 min, indicating that it can form stable structures post-printing. In the shear thinning test, as shown in fig. S1b), it was found that the viscosity of all the hydrogel samples was decreased with the increased shear rate, demonstrating a shear thinning behaviour. Besides, 1-5% exhibited the lowest viscosity among the entire shear rate, suggesting that they had a poor printability. Interestingly, the viscosity of 6-8% were higher, showing that chitosan acts a viscosity modifier to enhance printability. The viscosity of 8% was highest at low shear rate among all and did not flow after 10 min due to gravity, therefore the ink with 8% chitosan was the most suitable composition to form stable 3D structures.

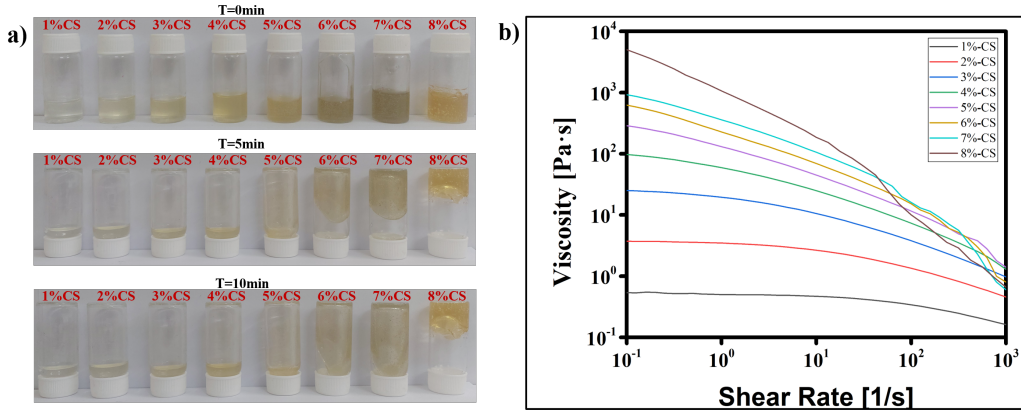


Figure S1: Printability of 1-8% chitosan hydrogel: a) flow behaviour of each hydrogel, b) viscosity as a function of shear rate at room temperature

2. Crosslinking procedure and mechanism

The samples were 3D printed using chitosan ink (the ink contains citric acid, but it's not yet cross-linked). All the samples were treated at 50°C in

22 a vacuum oven for 24 hours and the samples obtained were termed an un-
 23 crosslinked printed samples. To obtain cross-linked samples, the dried uncross-
 24 linked samples containing citric acid were treated at 150°C for 10 min in an
 25 oven for the crosslinking reaction to take place as shown in fig. S2 [1].

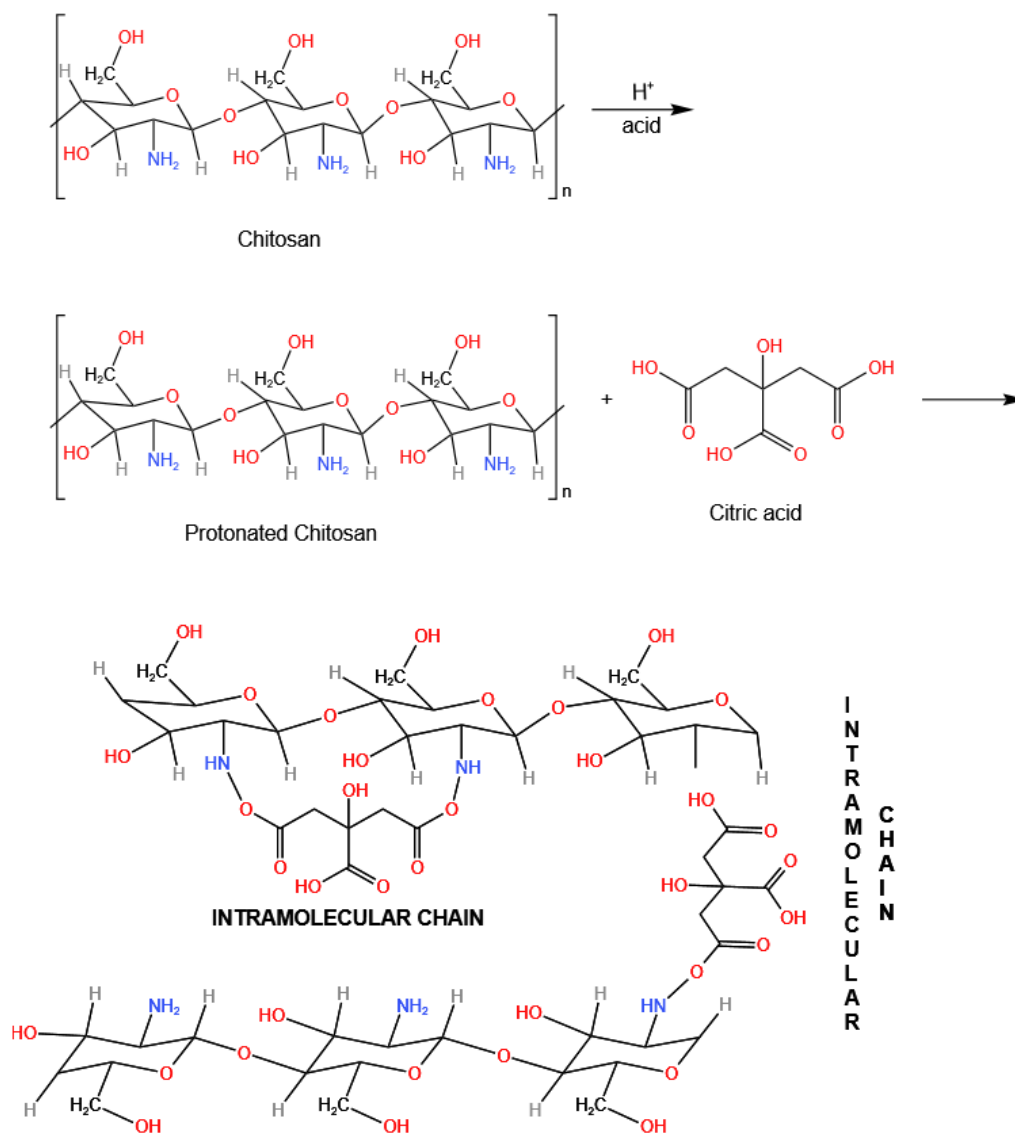


Figure S2: Crosslinking mechanism of chitosan with citric acid

To study the effect of crosslinking and mechanism of crosslinking, swelling measurements and FTIR analysis were conducted.

- **Swelling measurements**

Samples of diameter 20 mm were 3D printed for the swelling measurements. The swelling degree of the uncrosslinked, and cross-linked printed samples were measured by gravimetric method. Samples of each system were immersed into 100ml of distilled water. Weight of each sample was measured before (W_1) and after immersion (W_2) for 24 hours at room temperature. Swelling degree was calculated by the following equation:

$$S\% = (W_2 - W_1)/W_1 * 100 \quad (1)$$

Measurements were taken in triplicate for each system. The results showed that the uncrosslinked samples and cross-linked samples had swelling degree of 1550% and 250% respectively. The decreasing in the swelling values in the citric acid cross-linked structures could be explained by a higher crosslinking degree, which reduced the free volume and chain mobility, thus, hindering the water entry in the networks[2, 3].

- **Fourier Transform Infrared Spectrophotometer**

The FTIR spectra of chitosan samples were obtained by using Bruker's fourier transform infrared spectrophotometer. 64 scans were collected in the range 800-4000 cm^{-1} with a resolution of 8 cm^{-1} . The IR spectra were analysed using lab solutions series software.

Fig.S3 shows the ATR-FTIR spectrum of the chitosan(CS) and chitosan-citric acid(CS-CA) cross-linked films. The main characteristic bands associated with chitosan molecular were detected from the spectra. The amide *I* band is associated with stretching vibrations of C-O and appears at 1642 cm^{-1} . While, the amide *II* associated with C-N stretching and N-H vibration, has a band at 1577 cm^{-1} . The ATR-FTIR result implies that citric acid has been cross-linked with the chitosan molecules at 150°C for 10 min successfully.

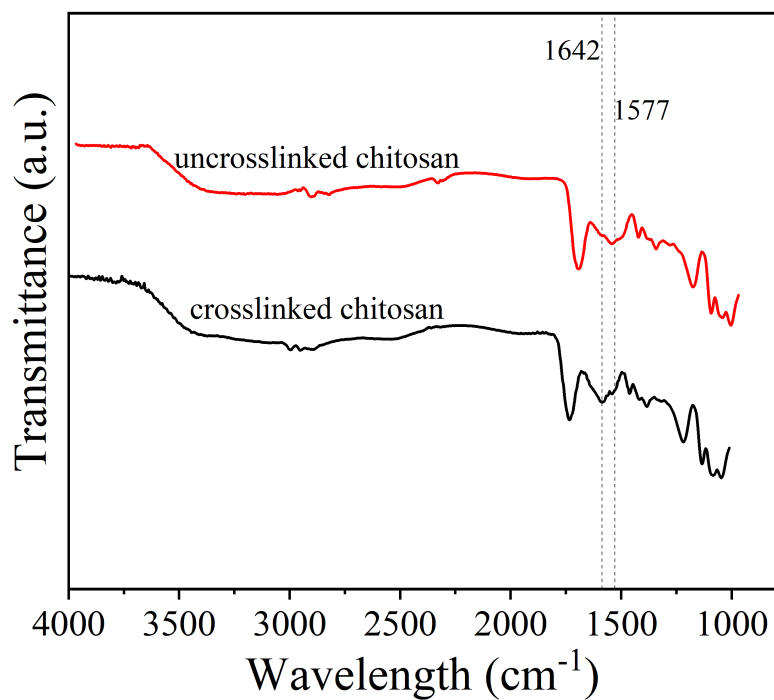


Figure S3: FTIR spectra of 3D printed chitosan films before and after the crosslinking with citric acid

57 3. Printability and shape fidelity

- 58 • Study of extrusion pressure (see figureS4)
- 59 • Study of printing speed (see figure S5)
- 60 • Analysis of shape fidelity (see figure S6)

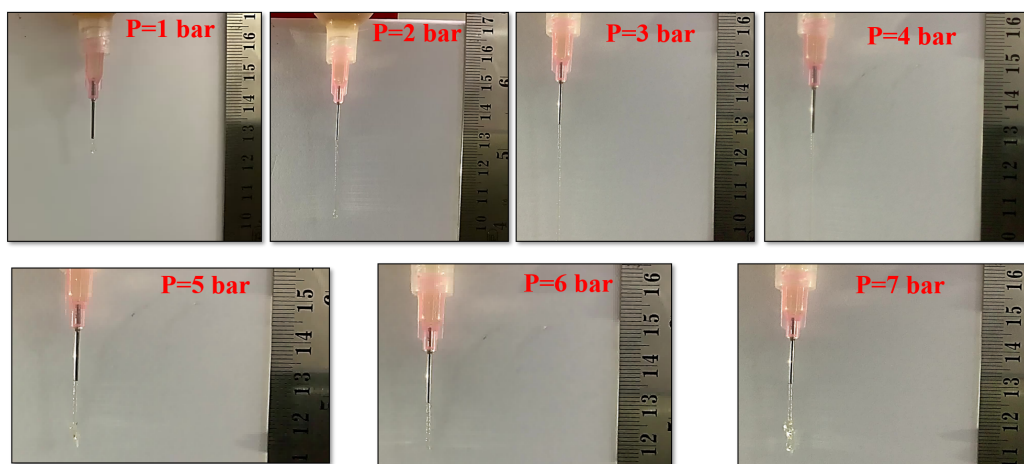


Figure S4: Study of extrusion pressure at different pressure(1-7 bar)

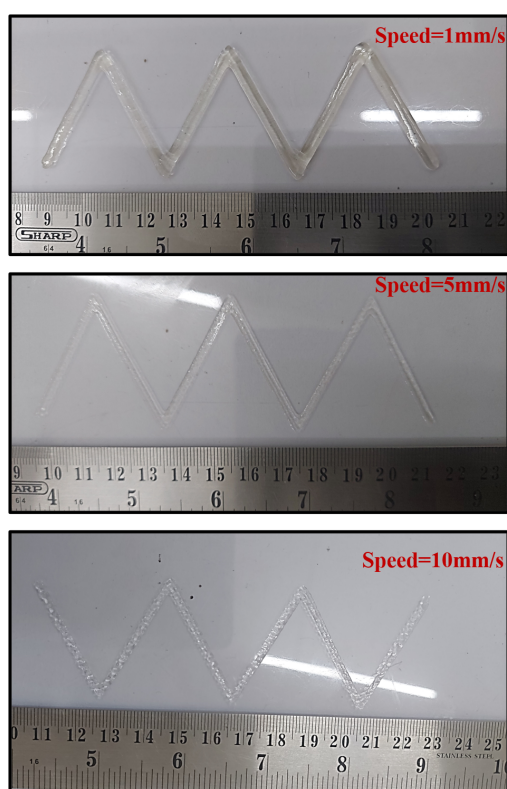
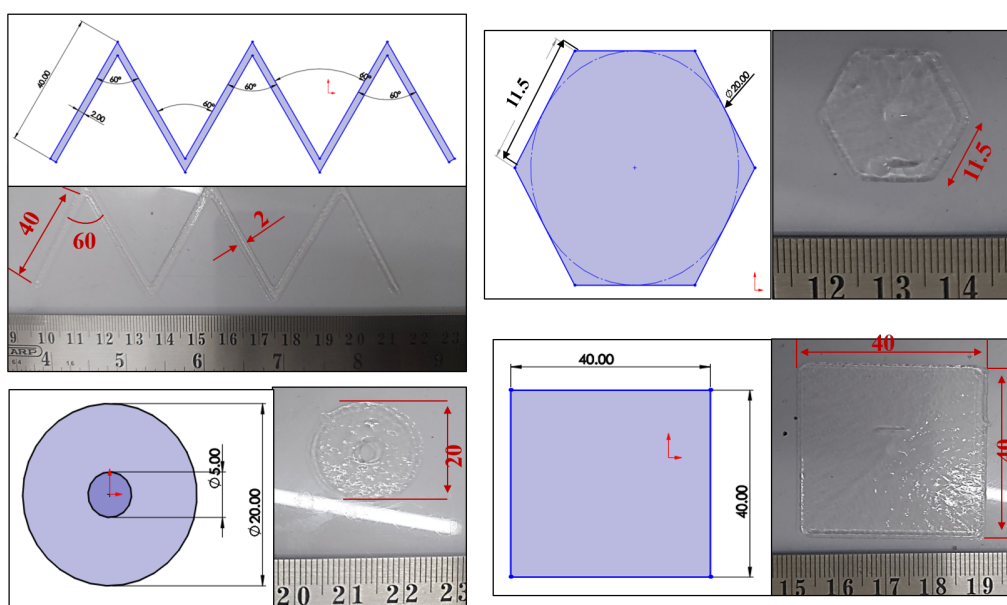


Figure S5: Study of printability at different printing speed(1-10 mm/s)



All dimensions are in mm

Figure S6: Comparison of designed part with printed structures using optimal printing parameters

61 **References**

- 62 [1] A. Awadhiya, D. Kumar, V. Verma, Crosslinking of agarose bioplastic
63 using citric acid, Carbohydrate Polymers 151 (2016) 60–67.
- 64 [2] M. Sheikhi, F. Rafiemanzelat, S. Ghodsi, L. Moroni, M. Setayeshmehr, 3d
65 printing of jammed self-supporting microgels with alternative mechanism
66 for shape fidelity, crosslinking and conductivity, Additive Manufacturing
67 58 (2022) 102997.
- 68 [3] P. Guerrero, A. Muxika, I. Zarandona, K. De La Caba, Crosslinking of
69 chitosan films processed by compression molding, Carbohydrate polymers
70 206 (2019) 820–826.